



## C<sub>3</sub> and C<sub>4</sub> photosynthesis models: An overview from the perspective of crop modelling

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### ABSTRACT

Nearly three decades ago Farquhar, von Caemmerer and Berry published a biochemical model for C<sub>3</sub> photosynthetic rates (the FvCB model). The model predicts net photosynthesis ( $A$ ) as the minimum of the Rubisco-limited rate of CO<sub>2</sub> assimilation ( $A_c$ ) and the electron transport-limited rate of CO<sub>2</sub> assimilation ( $A_j$ ). Given its simplicity and the growing availability of the required enzyme kinetic constants, the FvCB model has been used for a wide range of studies, from analysing underlying C<sub>3</sub> leaf biochemistry to predicting photosynthetic fluxes of ecosystems in response to global warming. However, surprisingly, this model has seen limited use in existing crop growth models. Here we highlight the elegance, simplicity, and robustness of this model. In the light of some uncertainties with photosynthetic electron transport pathways, a recently extended FvCB model to calculate  $A_j$  is summarized.

Applying the FvCB-type model in crop growth models for predicting leaf photosynthesis requires a stomatal conductance ( $g_s$ ) model to be incorporated, so that intercellular CO<sub>2</sub> concentration ( $C_i$ ) can be estimated. In recent years great emphasis has been put on the significant drawdown of Rubisco carboxylation-site CO<sub>2</sub> concentration ( $C_c$ ) relative to  $C_i$ . To account for this drawdown, mesophyll conductance ( $g_m$ ) for CO<sub>2</sub> transfer can be added. We present an analytical algorithm that incorporates a  $g_s$  model and uses  $g_m$  as a temperature-dependent parameter for calculating  $A$  under various environmental scenarios.

Finally we discuss a C<sub>4</sub>-equivalent version of the FvCB model. In addition to the algorithms already elaborated for C<sub>3</sub> photosynthesis, most important algorithms for C<sub>4</sub> photosynthesis are those that capture the CO<sub>2</sub> concentrating mechanism and the extra ATP requirement by the C<sub>4</sub> cycle. Although the current estimation of the C<sub>4</sub> enzyme kinetic constants is less certain, applying FvCB-type models to both C<sub>3</sub> and C<sub>4</sub> crops is recommended to accurately predict the response of crop photosynthesis to multiple, interactive environmental variables.

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### 1. Introduction

Photosynthesis is the primary physiological process that drives plant growth and crop productivity and influences many other plant processes. It is also strongly affected by environmental stresses. Its study becomes increasingly important in the context of assessing the impact of climate change on agro-ecosystem function and of exploring opportunities for bio-based energy production. Therefore, mechanistic quantification of photosynthesis deserves further attention in dynamic simulation models of crop growth (crop growth models hereafter).

Photosynthesis is one of the most studied and best understood physiological processes. In the wake of the advent of the systems

biology era, detailed biochemical models of photosynthesis, comprising its light reactions, electron and proton transport, enzymatic reactions, and regulatory functions, have been recently developed (e.g., [1,2]). These models are very useful for any transient, kinetic investigation of photosynthesis in response to environmental variables. Because of their complexity and required short time resolution, however, these models are unsuitable for use as a leaf-level model for large-scale modelling of photosynthesis of crop canopies or vegetation. The biochemical model for C<sub>3</sub> photosynthetic CO<sub>2</sub> assimilation published by Farquhar, von Caemmerer and Berry [3] (the FvCB model hereafter) makes no attempt to model all the processes of photosynthesis from light harvesting to metabolism but rather simplifies, summarizes and synthesizes the knowledge of the contributing mechanisms by focusing on a few key processes. Because of its excellent performance given its simplicity [4], and the growing availability of the required enzyme kinetic constants (e.g., [5,6]), the FvCB model has been used for a

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wide range of studies, from analysing underlying  $C_3$  leaf biochemistry (e.g., [4]) to predicting photosynthetic fluxes of ecosystems in response to global environmental change (e.g., [7,8]). However, this model has seen very limited use in crop growth models.

Our objective is to ‘catalyse’ the use of the FvCB model by the crop modelling community. To this end, we shall (1) high-

light the elegance, simplicity, and robustness of the model, (2) summarize an extended FvCB model that addresses some uncertainties with photosynthetic electron ( $e^-$ ) transport pathways, and (3) discuss the mathematical coupling of the FvCB model with  $CO_2$ -diffusional conductance models so that the value of  $A$  (see Table 1 for model variable definition) can be modelled for any

**Table 1**  
List of main variables used in the models and their units.

Variable	Definition	Unit
$a_1$	An empirical coefficient, see Eq. (15a)	–
$A$	Net photosynthesis rate	$\mu\text{mol CO}_2 \text{ m}^{-2} \text{ s}^{-1}$
$A_c$	Rubisco activity limited net photosynthesis rate	$\mu\text{mol CO}_2 \text{ m}^{-2} \text{ s}^{-1}$
$A_j$	Electron transport limited net photosynthesis rate	$\mu\text{mol CO}_2 \text{ m}^{-2} \text{ s}^{-1}$
$b_1$	An empirical coefficient, see Eq. (15a)	$\text{kPa}^{-1}$
$C_a$	Ambient air $CO_2$ partial pressure	$\mu\text{bar}$ or $\mu\text{mol mol}^{-1}$
$C_c$	Chloroplast $CO_2$ partial pressure	$\mu\text{bar}$
$C_i$	Intercellular $CO_2$ partial pressure	$\mu\text{bar}$
$C_{i^*}$	$C_i$ -based $CO_2$ compensation point in the absence of $R_d$	$\mu\text{bar}$
$C_s$	Leaf-surface $CO_2$ partial pressure	$\mu\text{bar}$
$C_{s^*}$	$C_s$ -based $CO_2$ compensation point in the absence of $R_d$	$\mu\text{bar}$
$D$	Deactivation energy	$\text{J mol}^{-1}$
$E$	Activation energy	$\text{J mol}^{-1}$
$f_{cyc}$	Fraction of electrons at PSI that follow cyclic transport around PSI	–
$f_{pseudo}$	Fraction of electrons at PSI that follow pseudocyclic transport	–
$f_Q$	Fraction of electrons at reduced plastoquinone that follow the Q-cycle	–
$f_{VPd}$	Factor for describing the effect of leaf-to-air vapour difference on $g_s$	–
$g_0$	Residual stomatal conductance when irradiance approaches zero	$\text{mol m}^{-2} \text{ s}^{-1} \text{ bar}^{-1}$
$g_b$	Boundary-layer conductance	$\text{mol m}^{-2} \text{ s}^{-1} \text{ bar}^{-1}$
$g_{bs}$	Bundle-sheath conductance	$\text{mol m}^{-2} \text{ s}^{-1} \text{ bar}^{-1}$
$g_m$	Mesophyll diffusion conductance	$\text{mol m}^{-2} \text{ s}^{-1} \text{ bar}^{-1}$
$g_s$	Stomatal conductance	$\text{mol m}^{-2} \text{ s}^{-1} \text{ bar}^{-1}$
$h$	Number of protons required to produce one ATP	$\text{mol mol}^{-1}$
$I_{abs}$	Photon flux density absorbed by leaf photosynthetic pigments	$\mu\text{mol photon m}^{-2} \text{ s}^{-1}$
$I_{inc}$	Photon flux density incident to leaves	$\mu\text{mol photon m}^{-2} \text{ s}^{-1}$
$J$	Rate of $e^-$ transport	$\mu\text{mol e}^- \text{ m}^{-2} \text{ s}^{-1}$
$J_2$	Rate of all $e^-$ transport through PSII	$\mu\text{mol e}^- \text{ m}^{-2} \text{ s}^{-1}$
$J_{max}$	Maximum value of $J$ under saturated light	$\mu\text{mol e}^- \text{ m}^{-2} \text{ s}^{-1}$
$J_{2max}$	Maximum value of $J_2$ under saturated light	$\mu\text{mol e}^- \text{ m}^{-2} \text{ s}^{-1}$
$k_p$	Initial carboxylation efficiency of the PEP carboxylase	$\text{mol m}^{-2} \text{ s}^{-1} \text{ bar}^{-1}$
$K_{mC}$	Michaelis–Menten constant of Rubisco for $CO_2$	$\mu\text{bar}$
$K_{mO}$	Michaelis–Menten constant of Rubisco for $O_2$	$\mu\text{bar}$
$L$	Leak rate of $CO_2$ out of the bundle sheath	$\mu\text{mol CO}_2 \text{ m}^{-2} \text{ s}^{-1}$
$O$	Oxygen partial pressure	$\mu\text{bar}$
$O_i$	Intercellular oxygen partial pressure	$\mu\text{bar}$
$O_{bs}$	Bundle-sheath oxygen partial pressure	$\mu\text{bar}$
$P_{max}$	Maximum gross photosynthetic rate under saturated irradiance	$\mu\text{mol CO}_2 \text{ m}^{-2} \text{ s}^{-1}$
$R$	Universal gas constant (=8.314)	$\text{J K}^{-1} \text{ mol}^{-1}$
$R_d$	Day respiration (respiratory $CO_2$ release other than by photorespiration)	$\mu\text{mol CO}_2 \text{ m}^{-2} \text{ s}^{-1}$
$R_m$	Day respiration in the mesophyll	$\mu\text{mol CO}_2 \text{ m}^{-2} \text{ s}^{-1}$
$s$	A lumped parameter, equal to $\rho_2 \beta [1 - f_{pseudo} / (1 - f_{cyc})]$	–
$S$	Entropy term	$\text{J K}^{-1} \text{ mol}^{-1}$
$S_{c/o}$	Relative $CO_2/O_2$ specificity factor for Rubisco	$\text{bar bar}^{-1}$
$T$	Leaf temperature	$^\circ\text{C}$
$V_{cmax}$	Maximum rate of Rubisco activity-limited carboxylation	$\mu\text{mol CO}_2 \text{ m}^{-2} \text{ s}^{-1}$
$V_p$	PEP carboxylation rate	$\mu\text{mol CO}_2 \text{ m}^{-2} \text{ s}^{-1}$
$V_p(J_2)$	$e^-$ transport-limited PEP carboxylation rate	$\mu\text{mol CO}_2 \text{ m}^{-2} \text{ s}^{-1}$
$V_{pmax}$	Maximum PEP carboxylation rate	$\mu\text{mol CO}_2 \text{ m}^{-2} \text{ s}^{-1}$
$x$	Fraction of PSII $e^-$ transport rate partitioned to the $C_4$ cycle	–
$z$	A lumped parameter, equal to $(2 + f_Q - f_{cyc}) / [h(1 - f_{cyc})]$	$\text{mol mol}^{-1}$
$\alpha$	Fraction of PSII activity in the bundle sheath	–
$\alpha_{(LL)}$	Conversion efficiency of absorbed light into $J$ at strictly limiting light	$\text{mol e}^- (\text{mol photon})^{-1}$
$\alpha_{2(LL)}$	Quantum efficiency of PSII $e^-$ transport under strictly limiting light, on the combined PSI- and PSII-absorbed light basis	$\text{mol e}^- (\text{mol photon})^{-1}$
$\beta$	Absorptance by leaf photosynthetic pigments	–
$\gamma^*$	Half of the reciprocal of $S_{c/o}$	$\text{bar bar}^{-1}$
$\kappa_{2(LL)}$	Conversion efficiency of incident light into $J$ at strictly limiting light	$\text{mol e}^- (\text{mol photon})^{-1}$
$\theta$	Convexity factor for response of $J$ to irradiance	–
$\theta_2$	Convexity factor for response of $J_2$ to absorbed light	–
$\rho_2$	Proportion of absorbed light partitioned to PSII	–
$\Phi_{1(LL)}$	Quantum efficiency of PSI $e^-$ flow at the strictly limiting light level	$\text{mol e}^- (\text{mol photon})^{-1}$
$\Phi_2$	Quantum efficiency of PSII $e^-$ flow on PSII-absorbed light basis	$\text{mol e}^- (\text{mol photon})^{-1}$
$\Phi_{2(LL)}$	Value of $\Phi_2$ at the strictly limiting light level	$\text{mol e}^- (\text{mol photon})^{-1}$
$\Phi_{CO_2(LL)}$	Quantum efficiency of $CO_2$ assimilation at strictly limiting light	$\text{mol CO}_2 (\text{mol photon})^{-1}$
$\Gamma$	$C_c$ - or $C_i$ -based $CO_2$ compensation point in the presence of $R_d$	$\mu\text{bar}$
$\Gamma^*$	$C_c$ -based $CO_2$ compensation point in the absence of $R_d$	$\mu\text{bar}$

environmental scenario. Finally, we shall describe and discuss an equivalent, combined model for  $C_4$  photosynthesis and conductance, building upon previous  $C_4$  photosynthesis models, e.g., as described by von Caemmerer and Furbank [9].

## 2. The Farquhar, von Caemmerer and Berry model

The FvCB model predicts  $A$  as the minimum of the Rubisco-limited rate of  $CO_2$  assimilation ( $A_c$ ) and the electron transport-limited rate of  $CO_2$  assimilation ( $A_j$ ):

$$A = \min(A_c, A_j) \quad (1)$$

An illustration of the two parts of limitations along the  $CO_2$ -response curves is given in Fig. 1. Sharkey et al. [10] have drawn attention to a third limitation by triose phosphate utilization, which is not discussed here because it comes into play only occasionally at very high  $CO_2$  levels.

The value of  $A_c$  is calculated as a function of the maximum carboxylation capacity of Rubisco ( $V_{cmax}$ ) by:

$$A_c = \frac{(C_c - \Gamma^*)V_{cmax}}{C_c + K_{mC}(1 + O/K_{mO})} - R_d \quad (2)$$

where  $C_c$  is the  $CO_2$  partial pressure at the carboxylating sites of Rubisco,  $K_{mC}$  and  $K_{mO}$  are Michaelis–Menten constants of Rubisco for  $CO_2$  and  $O_2$ , respectively, and  $\Gamma^*$  is the  $CO_2$  compensation point in the absence of day respiration ( $R_d$ ).

In the calculation of  $A_j$ , the FvCB model assumes 100% non-cyclic  $e^-$  transport, thus excluding cyclic  $e^-$  transport around PSI (CET). There are two widely used forms of the equation for electron transport-limited rate of photosynthesis:

$$A_j = \frac{(C_c - \Gamma^*)J}{4C_c + 8\Gamma^*} - R_d \quad (3a)$$

$$A_j = \frac{(C_c - \Gamma^*)J}{4.5C_c + 10.5\Gamma^*} - R_d \quad (3b)$$

The relationship between  $e^-$  transport rate ( $J$ ) in Eqs. (3a), (3b) and irradiance was first described as a rectangular hyperbola [11], using quantum yield of  $e^-$  transport under limiting light ( $\alpha_{(LL)}$ ) and the maximum capacity of  $e^-$  transport ( $J_{max}$ ). Following Farquhar and Wong [12], most applications of the FvCB model, however,

describe  $J$  as a non-rectangular hyperbolic function of irradiance by:

$$J = \frac{\left( \alpha_{(LL)}I_{abs} + J_{max} - \sqrt{(\alpha_{(LL)}I_{abs} + J_{max})^2 - 4\theta J_{max}\alpha_{(LL)}I_{abs}} \right)}{2\theta} \quad (4)$$

where  $\theta$  is the convexity of the response curve of  $J$  to light absorbed by photosynthetic pigments ( $I_{abs}$ ). Equations like Eq. (4) that describe the light response of  $e^-$  transport rate mimic well the photosynthetic down-regulation induced by high light levels via mechanisms such as non-photochemical quenching and chloroplast avoidance movement [13]. The theoretical maximum value for  $\alpha_{(LL)}$  is 0.5 mol electron per mol photon absorbed [3] because one quantum must be absorbed by each of the two photosystems to move an electron from the level of  $H_2O$  to that of  $NADP^+$ . However, in actual applications (e.g., [14–17]),  $\alpha_{(LL)}$  has been empirically adjusted to a lower value to agree with a measured quantum efficiency for  $CO_2$  uptake that is often lower than that expected from the theoretical maximum.

The temperature dependence of  $R_d$  and kinetic properties of Rubisco (involving three parameters  $V_{cmax}$ ,  $K_{mC}$  and  $K_{mO}$ ) in Eq. (2) is described by an Arrhenius function normalized with respect to their values at 25 °C:

$$\text{Parameter} = \text{Parameter}_{25} e^{(T-25)E/[298R(T+273)]} \quad (5)$$

where  $T$  is leaf temperature;  $E$  is the activation energy, defining the responsiveness of the relevant parameter to temperature;  $R$  is the universal gas constant. A modified Arrhenius function is used to describe the optimum response of other parameters (e.g.,  $J_{max}$ ) to temperature as [17]:

$$\text{Parameter} = \text{Parameter}_{25} e^{(T-25)E/[298R(T+273)]} \times \frac{1 + e^{(298S-D)/(298R)}}{1 + e^{(T+273)S-D/[R(T+273)]}} \quad (6)$$

where  $S$  is an entropy term;  $E$  and  $D$  are the energies of activation and deactivation, defining the responsive shape of the sub- and supra-optimal ranges, respectively. June et al. [18] described an alternative, simpler equation for this optimum response. It may be argued that these response equations lack a mechanistic basis since parameters such as  $E$  are often estimated from fitting to experimental data [17,19]. However, the equations do provide a flexibility to accommodate possible variability in the shape of the temperature response, e.g., among genotypes or species.

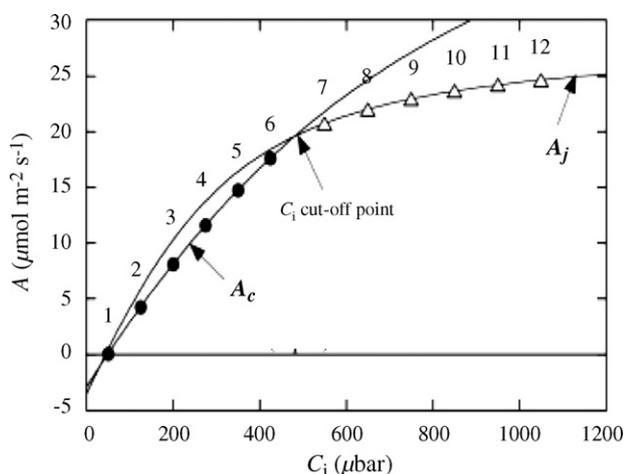
The value  $\Gamma^*$  depends on the  $O_2$  concentration ( $O$ ) and the Rubisco  $CO_2/O_2$  specificity factor ( $S_{c/o}$ ) as follows:

$$\Gamma^* = \frac{0.5O}{S_{c/o}} \quad (7)$$

where the factor 0.5 is mol  $CO_2$  released when Rubisco catalyses the reaction with 1 mol  $O_2$  in photorespiration [3]. The value of  $S_{c/o}$ , which also depends on temperature and can be described by Eq. (5) in which  $E$  should be negative to account for a decline of the Rubisco affinity for  $CO_2$  with increasing temperature.

The values of  $V_{cmax}$  and  $J_{max}$  depend on the concentration of relevant enzymes. For practical purposes,  $V_{cmax25}$  and  $J_{max25}$  increase linearly with leaf nitrogen content (e.g., [14,20]). Other parameters are assumed constant, and their *in vivo* estimates are shown in Table 2.

The basic equations of the FvCB model, Eqs. (1)–(4), capture the response of  $C_3$  photosynthesis to irradiance,  $CO_2$  and  $O_2$  levels, although the model may not predict often observed increases in the initial slope of the photosynthetic response curve to  $CO_2$  with increased light levels (e.g., [21,22]). Coupled with auxiliary equations, Eqs. (5)–(7) and the linear relation between leaf nitrogen level and  $V_{cmax25}$  or  $J_{max25}$  (e.g., [14,16,23,24]), the model also



**Fig. 1.** An idealized curve for the response of net  $CO_2$  assimilation rate ( $A$ ) in  $C_3$  plants to intercellular  $CO_2$  partial pressure ( $C_i$ ), in which 12 data points are shown. Points 1–6 locate within the range of the Rubisco-limited rate ( $A_c$ ) whereas points 7–12 are within the range of electron transport-limited rate ( $A_j$ ). The portions of each curve without data points are the extended parts as given by the  $A_c$  and  $A_j$  equation, respectively. The minimum of  $A_c$  and  $A_j$  gives the modelled  $CO_2$  response curve as indicated by the 12 data points.

**Table 2**  
Indicative values for constants used in the C<sub>3</sub> and C<sub>4</sub> photosynthesis models.

Constant	Unit	C <sub>3</sub>		C <sub>4</sub>	
		Value	Reference	Value	Reference
S <sub>c/o25</sub>	bar bar <sup>-1</sup>	2800	Bernacchi et al. [6]	2590	von Caemmerer [54]
K <sub>mC25</sub>	μbar	270	Bernacchi et al. [6]	650	von Caemmerer [54]
K <sub>mO25</sub>	μbar	165000	Bernacchi et al. [6]	450000	von Caemmerer [54]
E <sub>Sc/o</sub>	J mol <sup>-1</sup>	-24460	Bernacchi et al. [6]	?	
E <sub>Vcmax</sub>	J mol <sup>-1</sup>	65330	Bernacchi et al. [71]	67300	Massad et al. [53]
E <sub>KmC</sub>	J mol <sup>-1</sup>	80990	Bernacchi et al. [6]	?	
E <sub>KmO</sub>	J mol <sup>-1</sup>	23720	Bernacchi et al. [6]	?	
E <sub>Rd</sub>	J mol <sup>-1</sup>	46390	Bernacchi et al. [71]	?	
E <sub>Jmax</sub>	J mol <sup>-1</sup>	26900–94400	Yin et al. [19]	77900	Massad et al. [53]
D <sub>Jmax</sub>	J mol <sup>-1</sup>	200000	Medlyn et al. [17]	192000	Massad et al. [53]
S <sub>Jmax</sub>	JK <sup>-1</sup> mol <sup>-1</sup>	650	Harley et al. [14]	630	Massad et al. [53]
E <sub>Vpmax</sub>	J mol <sup>-1</sup>	na		70400	Massad et al. [53]
D <sub>Vpmax</sub>	J mol <sup>-1</sup>	na		118000	Massad et al. [53]
S <sub>Vpmax</sub>	JK <sup>-1</sup> mol <sup>-1</sup>	na		380	Massad et al. [53]
E <sub>gm</sub>	J mol <sup>-1</sup>	49600	Bernacchi et al. [6]	na	
D <sub>gm</sub>	J mol <sup>-1</sup>	437400	Bernacchi et al. [6]	na	
S <sub>gm</sub>	JK <sup>-1</sup> mol <sup>-1</sup>	1400	Bernacchi et al. [6]	na	
k <sub>p</sub>	mol m <sup>-2</sup> s <sup>-1</sup> bar <sup>-1</sup>	na		0.7	Collatz et al. [51]
θ, or θ <sub>2</sub>	–	0.7 or variable	von Caemmerer [54] Bernacchi et al. [38]	0.7	von Caemmerer [54]
s	–	0.33–0.41	Yin et al. [20]	na	
Φ <sub>1(LL)</sub>	mol mol <sup>-1</sup>	0.95–1.0	Trissl and Wilhelm [36]	0.95–1.0	Kingston-Smith et al. [72]
Φ <sub>2(LL)</sub>	mol mol <sup>-1</sup>	≈ 0.75	Yin et al. [20]	≈ 0.75	Assumed as C <sub>3</sub> value
β	–	0.84 or variable	Standard value in Li-Cor, Evans [73]	0.84	Standard value in Li-Cor
h	mol mol <sup>-1</sup>	3, 4, 14/3	Yin et al. [70]	3 or 4	von Caemmerer [54]
x	–	na		0.4	von Caemmerer [54]
f <sub>pseudo</sub>	–	≈ 0.1	Yin et al. [35]	≈ 0.1	Laisk and Edwards [59]
f <sub>Q</sub>	–	0–1	von Caemmerer [54]	1	Furbank et al. [56]
a <sub>1</sub>	–	≈ 0.85	†	≈ 0.85	†
b <sub>1</sub>	kPa <sup>-1</sup>	≈ 0.14	†	≈ 0.20	†

?: not known to the authors; and their values for C<sub>3</sub> leaves are used tentatively here for simulation. na: not applicable. (†) Derived here from the data of Morison and Gifford [74].

quantifies the photosynthetic responses to temperature and nitrogen level. Putting all together, the responses of photosynthesis to these environmental variables can be unambiguously predicted mechanistically by the model.

CO<sub>2</sub> exchange at the leaf level can now be measured routinely with commercially available equipment [25] such as the Li-Cor 6400 (Li-Cor Inc., Lincoln, NE, USA). With such measurements, parameters of the FvCB model can be estimated using nonlinear regression fitting (e.g., [14,17,19,26–28]), especially when CO<sub>2</sub> exchange measurements are combined with chlorophyll fluorescence measurements [20].

Leaf photosynthesis in crop growth simulation has traditionally been calculated using a family of empirical equations for its light response curves. Typically these curves are characterized by two parameters: initial quantum use efficiency of CO<sub>2</sub> assimilation at limiting lights (Φ<sub>CO<sub>2</sub>(LL)</sub>) and maximum gross CO<sub>2</sub> assimilation rate under a saturating light (P<sub>max</sub>). Empirical calibration procedures are needed if the models are used to predict the effect of environmental variables other than light on photosynthesis. According to van Oijen et al. [29], both Φ<sub>CO<sub>2</sub>(LL)</sub> and P<sub>max</sub> can be formulated from the FvCB model:

$$\Phi_{CO_2(LL)} = \frac{\alpha_{(LL)}(C_c - \Gamma_*)}{4.5C_c + 10.5\Gamma_*} \tag{8}$$

$$P_{max} = \frac{(C_c - \Gamma_*)V_{c\ max}}{C_c + K_{mC}(1 + O/K_{mO})} \tag{9}$$

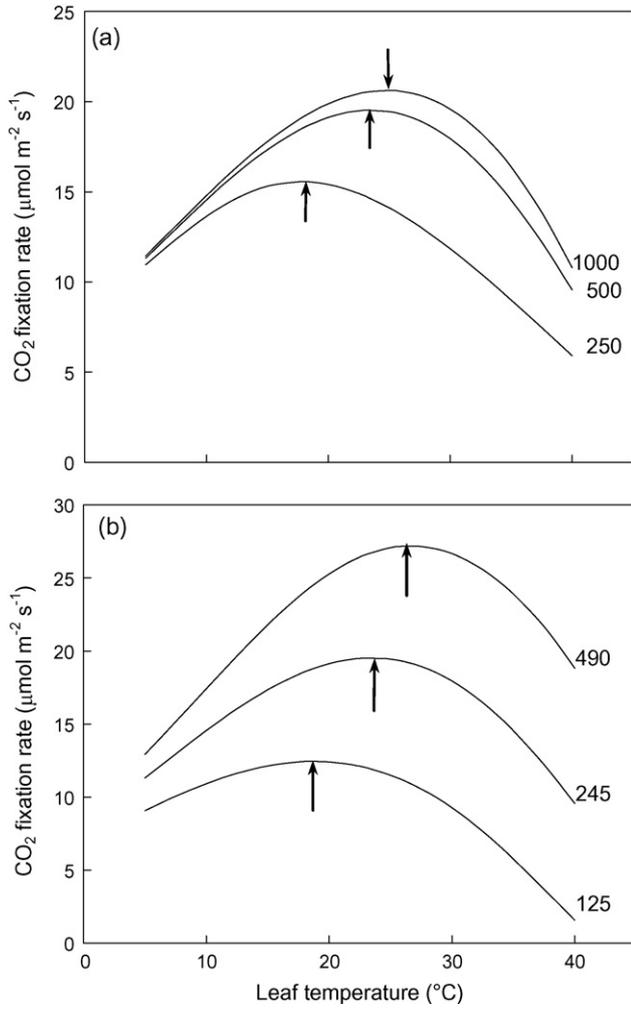
where Eq. (8) holds if Eq. (3b) applies; if Eq. (3a) applies, the coefficients 4.5 and 10.5 should be replaced by 4 and 8, respectively. Eq. (9) holds in general, but according to the FvCB model, P<sub>max</sub> can sometimes be determined as (A<sub>j</sub> + R<sub>d</sub>) rather than (A<sub>c</sub> + R<sub>d</sub>), as done by Boote and Pickering [30]. When Eqs. (8) and (9) are used there

is no need to conduct empirical model calibrations or corrections for a change in environmental conditions such as CO<sub>2</sub> concentration. Note that empirical calibrations were implemented in some crop growth models, e.g., the rice model ORYZA ([31]; see a later section).

Observed temperature response curves of photosynthesis under various CO<sub>2</sub> or light levels (e.g., [32]) provide strong support for the FvCB model, although uncertainties may exist for the modelled response over the range of low temperatures down to 0 °C. This model predicts that at high CO<sub>2</sub> levels the rate of photorespiration is reduced, thereby extending the temperature range where CO<sub>2</sub> assimilation rate is positive. The model also predicts an increasing and more pronounced temperature optimum with increasing either CO<sub>2</sub> or light levels (Fig. 2). Such a shift in the temperature optimum with a change in other environmental factors is critically important for correctly assessing the impact of climate change [7], e.g., on crop production and agro-ecosystem functioning. However, such interactions of temperature with light or CO<sub>2</sub> levels cannot be predicted by any empirical photosynthesis models using a simple light response equation.

### 3. An extended model and its use for calibrating the Farquhar, von Caemmerer and Berry model

There is, however, an ambiguity in the FvCB model for calculating A<sub>j</sub>, as shown by the use of two equations—Eqs. (3a) and (3b). In applying the FvCB model, some researchers (e.g., [14]) have used Eq. (3a), whereas others (e.g., [7]) used Eq. (3b), with little explanation why one form was preferred over the other. Eqs. (3a) and (3b) were derived by assuming that the noncyclic e<sup>-</sup> transport was the only photosynthetic e<sup>-</sup> transport process active in leaves. This assumption results in two possible outcomes: either the NADPH or the ATP



**Fig. 2.** An example of response curves of net CO<sub>2</sub> assimilation rate in a C<sub>3</sub> plant to leaf temperature, modelled using the FvCB model, (a) at three levels of irradiance (μmol m<sup>-2</sup> s<sup>-1</sup>) when C<sub>c</sub> was fixed at 245 μbar, and (b) at three levels of C<sub>c</sub> (μbar) when irradiance was fixed at 500 μmol m<sup>-2</sup> s<sup>-1</sup>. Arrows indicate the position of the optimum temperature.

supply will limit overall photosynthesis, applying to Eq. (3a) and Eq. (3b), respectively. Furthermore, Eq. (3a) assumes 100% linear e<sup>-</sup> transport (LET), the noncyclic e<sup>-</sup> flux used for carbon reduction and photorespiration. Eq. (3b) implies that in addition to the LET there is some pseudocyclic e<sup>-</sup> transport (PET), the noncyclic e<sup>-</sup> flux not used for carbon reduction and photorespiration. Comparing Eqs. (3a) and (3b) indicates that ATP production is more limiting than NADPH supply. Thus, in Eq. (3b), PET is assumed to occur *in vivo* to reduce or eliminate the ATP deficit.

Many studies (e.g., [14,15]) in which the FvCB model was applied calculated e<sup>-</sup> transport rate (*J*) using a constant quantum yield of e<sup>-</sup> transport, which is usually corrected empirically for the observed  $\Phi_{\text{CO}_2(\text{LL})}$ . In fact,  $\Phi_{\text{CO}_2(\text{LL})}$  is a function of excitation partitioning of absorbed light between photosystem I (PSI) and photosystem II (PSII), and e<sup>-</sup> transfer efficiencies of PSI and PSII. Furthermore, any involvement of CET will reduce observed  $\Phi_{\text{CO}_2(\text{LL})}$ . To account for the fraction of CET (*f<sub>cyc</sub>*) and PET (*f<sub>pseudo</sub>*) (Fig. 3) as well as for the difference between PSI and PSII e<sup>-</sup> transport efficiencies under limiting light ( $\Phi_{1(\text{LL})}$  and  $\Phi_{2(\text{LL})}$ , respectively) and for an uncertainty with regard to the operation of the Q-cycle (*f<sub>Q</sub>*) and the protons required for synthesising one mol ATP (*h*), the FvCB model for *A<sub>j</sub>* was extended analytically for a generalized stoichiometry ([19,35])

by:

$$A_j = J_2 \left( 1 - \frac{f_{\text{pseudo}}}{1 - f_{\text{cyc}}} \right) \frac{C_c - \Gamma_*}{4C_c + 8\Gamma_*} - R_d \quad (10a)$$

$$J_2 = \frac{\left( \alpha_{2(\text{LL})} I_{\text{abs}} + J_{2\text{max}} - \sqrt{(\alpha_{2(\text{LL})} I_{\text{abs}} + J_{2\text{max}})^2 - 4\theta_2 J_{2\text{max}} \alpha_{2(\text{LL})} I_{\text{abs}}} \right)}{2\theta_2} \quad (10b)$$

$$\alpha_{2(\text{LL})} = \frac{\Phi_{2(\text{LL})}(1 - f_{\text{cyc}})}{\Phi_{2(\text{LL})}/\Phi_{1(\text{LL})} + (1 - f_{\text{cyc}})} \quad (10c)$$

$$1 - f_{\text{cyc}} - f_{\text{pseudo}} = \frac{(4C_c + 8\Gamma_*)(2 + f_Q - f_{\text{cyc}})}{h(3C_c + 7\Gamma_*)} \quad (10d)$$

where *J<sub>2</sub>* is e<sup>-</sup> transport rate through PSII while CET is running simultaneously,  $\theta_2$  is the convexity of the response curve of *J<sub>2</sub>* to *I<sub>abs</sub>*, and *J<sub>2max</sub>* is the maximum capacity of *J<sub>2</sub>* under a saturating *I<sub>abs</sub>*. Eq. (10d) sets the requirement for the relation between *f<sub>cyc</sub>*, *f<sub>pseudo</sub>* and *f<sub>Q</sub>* if ATP and NADPH produced in the thylakoid reactions are to match the requirement by the carbon reduction cycle and photorespiration. As shown by Yin et al. [19], Eqs. (3a) and (3b) of the FvCB model are special cases of the extended model assuming the lack of CET (i.e., *f<sub>cyc</sub>* = 0) and an *h* of 3, with either no (Eq. (3b)) or only a small fraction (Eq. (3a)) of the electrons following the Q-cycle. Contemporary literature (e.g., [33,34]) indicates an *h* of 4 or even higher, coupled with the absolute operation of the Q-cycle (i.e., *f<sub>Q</sub>* = 1).

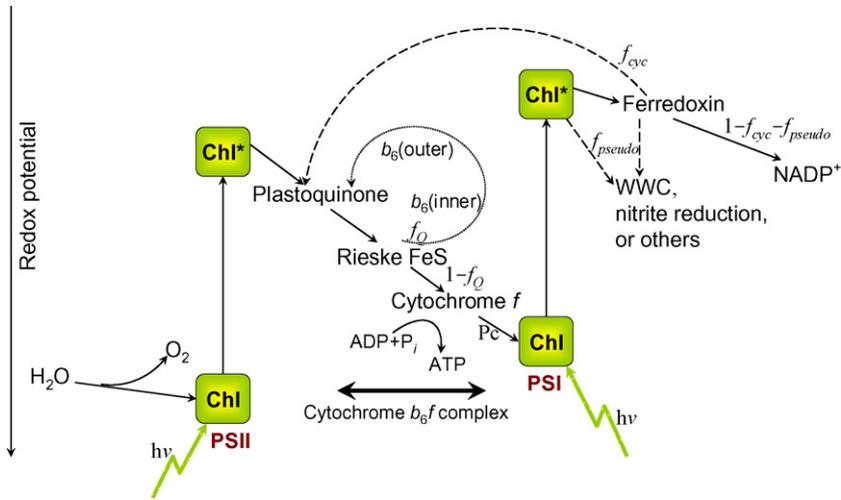
Yin et al. [35] discussed how the extended model can be used to theoretically infer the possible range of variation for values of *f<sub>cyc</sub>* and *f<sub>pseudo</sub>* based on gas exchange measurements on  $\Phi_{\text{CO}_2(\text{LL})}$  and biophysical measurements on  $\Phi_{1(\text{LL})}$  and  $\Phi_{2(\text{LL})}$ . The model can also calculate theoretically the value of  $\rho_2$ , the fraction of absorbed light partitioned to PSII, conditional to *f<sub>cyc</sub>* and PSI and PSII e<sup>-</sup> transport efficiencies. Furthermore, parameter  $\alpha_{(\text{LL})}$  in the FvCB model may be calculated as  $\Phi_{2(\text{LL})}/(1 + \Phi_{2(\text{LL})}/\Phi_{1(\text{LL})})$ , derived from Eq. (10c) for the case in the absence of CET. This allows  $\alpha_{(\text{LL})}$  in the FvCB model to be calculated from biophysical measurements for a difference in e<sup>-</sup> transport efficiency between PSI and PSII. The absolute maximum efficiency of PSI e<sup>-</sup> transport is 0.95 or greater [36]. Chlorophyll fluorescence measurements showed that the maximum PSII photochemical efficiency for dark-adapted leaves was quite conservative among plant species, about 0.83 [37]. However, Bernacchi et al. [38] and Yin et al. [20] have shown that the equivalent efficiency for leaves adapted to strictly limiting light is somewhat lower, ca. 0.75. Setting  $\Phi_{1(\text{LL})} = 1$  and  $\Phi_{2(\text{LL})} = 0.75$  gives  $\alpha_{(\text{LL})}$  in the FvCB model as  $\Phi_{2(\text{LL})}/(1 + \Phi_{2(\text{LL})}/\Phi_{1(\text{LL})}) = 0.43$ . This potentially avoids an empirical calibration for parameter  $\alpha_{(\text{LL})}$  in applying the FvCB model.

However, the estimation of  $\alpha_{(\text{LL})}$  for the FvCB model in that way is correct only if alternative e<sup>-</sup> transports (PET and CET) do not occur, which is highly unlikely in leaves. By comparing Eq. (3a) with Eq. (10a), the following equation can be derived:

$$J = J_2 \left( 1 - \frac{f_{\text{pseudo}}}{1 - f_{\text{cyc}}} \right) = \rho_2 \beta I_{\text{inc}} \Phi_2 \left( 1 - \frac{f_{\text{pseudo}}}{1 - f_{\text{cyc}}} \right) = s I_{\text{inc}} \Phi_2 \quad (11)$$

where  $s = \rho_2 \beta [1 - f_{\text{pseudo}}/(1 - f_{\text{cyc}})]$ , with  $\beta$  being the proportion of incident light absorbed by photosynthetic pigments. Although values of  $\beta$ ,  $\rho_2$ , *f<sub>cyc</sub>* and *f<sub>pseudo</sub>* may be variable, they can be practically assumed independent of light levels, as done in many physiological studies. So the lumped parameter *s* can also be considered as constant. Substituting Eq. (11) into Eq. (3a) gives:

$$A_j = \frac{(C_c - \Gamma_*) s I_{\text{inc}} \Phi_2}{4C_c + 8\Gamma_*} - R_d \quad (12)$$



**Fig. 3.** The Z scheme for photosynthetic thylakoid reactions showing linear (solid arrows), cyclic and pseudocyclic (dashed arrows) electron transport routes. From reduced ferredoxin, a fraction,  $f_{cyc}$ , of the electrons follows the cyclic mode around PSI. Another fraction,  $f_{pseudo}$ , of the electrons that have passed PSI follows the pseudocyclic mode for supporting processes such as the water–water cycle (WWC, see [69]), or nitrite reduction, or other minor metabolic processes. The remaining fraction,  $1 - f_{cyc} - f_{pseudo}$ , is transferred to NADP<sup>+</sup> –the terminal acceptor of the linear electron transport for generating NADPH in support of CO<sub>2</sub> reduction or photorespiration. The efficiency of ATP synthesis along the chain depends on the operation of the Q-cycle. The scheme shows that a fraction,  $f_Q$ , of the electrons followed the Q-cycle (dotted arrow) through the concerted action of the Rieske FeS and  $b_6$  of the cytochrome  $b_6f$  complex, and the remaining fraction,  $1 - f_Q$ , is transferred directly towards plastocyanin (Pc). Chl, chlorophyll;  $h\nu$ , photons absorbed either by PSI or by PSII. Redrawn from Yin et al. [70].

Combined measurements of CO<sub>2</sub> exchange and chlorophyll fluorescence over the range (e.g., low light and high CO<sub>2</sub> levels) where  $A$  is limited by  $A_j$ , could be used to determine the value of  $s$ , that is, as the slope of linear regression, based on Eq. (12), between  $A$  and  $I_{inc} \Phi_2/4$ , obtained under the non-photorespiratory condition (low O<sub>2</sub> and/or high CO<sub>2</sub>), at which  $\Gamma^*$  can be practically set to zero [20].

If parameter  $s$  is known, the efficiency for converting  $I_{inc}$  into LET under the strictly limiting light condition,  $\kappa_{2(LL)}$ , is given by:

$$\kappa_{2(LL)} = s \Phi_{2(LL)} \quad (13)$$

In analogy to Eq. (4), the equation for calculating  $J$  but as a function of incident irradiance can then be established as:

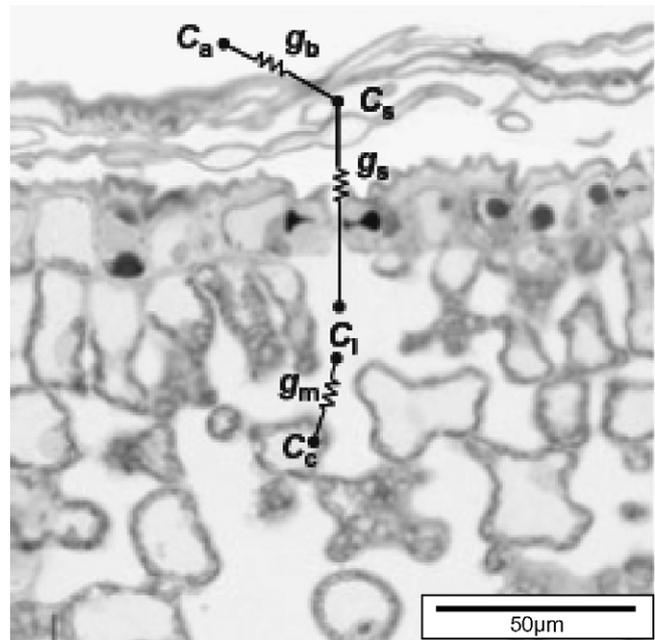
$$J = \frac{\left( \kappa_{2(LL)} I_{inc} + J_{max} - \sqrt{(\kappa_{2(LL)} I_{inc} + J_{max})^2 - 4\theta J_{max} \kappa_{2(LL)} I_{inc}} \right)}{2\theta} \quad (14)$$

Then,  $A_j$  could still be calculated by Eq. (3a), with  $J$  being given by Eq. (14). This way of calibration accounts for any occurrence of alternative e<sup>-</sup> transport. An advantage of using Eq. (14) with parameter  $\kappa_{2(LL)}$  over Eq. (4) is that parameter  $\kappa_{2(LL)}$  can be determined from combined measurements of gas exchange and chlorophyll fluorescence without the necessity to measure parameter  $\beta$ , which is very hard to measure and is often approximated by total leaf absorptance (note that there is probably significant absorptance by non-photosynthetic pigments in leaves). With this calibration procedure, only a single lumped parameter  $\kappa_{2(LL)}$  is needed, obviating the need for knowing individual underlying parameters  $\beta$ ,  $\rho_2$ ,  $f_{cyc}$  and  $f_{pseudo}$ , which would need more detailed measurements to estimate specifically. Yin et al. [20] showed for wheat leaves, that  $\kappa_{2(LL)}$  can be related to leaf nitrogen (N) content (in g m<sup>-2</sup>) as:  $\kappa_{2(LL)} = 0.2048 + 0.0435N$ . So, if no chlorophyll fluorescence or non-photorespiratory measurements were conducted for performing the aforementioned calibration procedure,  $\kappa_{2(LL)}$  can be first derived practically using this equation.

**4. Coupled modelling of C<sub>3</sub> photosynthesis and diffusional conductance**

The FvCB-type models, in principle, require  $C_c$  to be known *a priori*, although Farquhar et al. [3] initially used the intercellular

CO<sub>2</sub> level ( $C_i$ ) in places of  $C_c$  of Eqs. (2) and (3a), (3b). Diffusional conductance (including boundary-layer, stomatal and mesophyll components) is involved along the path of transfer from ambient CO<sub>2</sub> level ( $C_a$ ) to  $C_c$  (Fig. 4). The first two components determine the drawdown of  $C_i$  relative to  $C_a$ . Of the three components, stomatal conductance ( $g_s$ ) was formerly considered as most important, so in applying the FvCB model,  $C_i$  was then being treated as equal to  $C_c$  (e.g., [16]). In recent years, mesophyll conductance ( $g_m$ ) for CO<sub>2</sub> transfer has increasingly been found to be small enough for the existence of a significant drawdown of  $C_c$  relative to  $C_i$  (see the review of [39]). As it is the level of  $C_c$  rather than  $C_i$  that together



**Fig. 4.** Micrograph of the abaxial surface of a typical leaf, illustrating the pathway of CO<sub>2</sub> transfer from ambient air ( $C_a$ ) through leaf surface ( $C_s$ ) and intercellular air spaces ( $C_i$ ) to the Rubisco carboxylation-sites in chloroplasts ( $C_c$ ). Boundary-layer conductance ( $g_b$ ), stomatal conductance ( $g_s$ ), and mesophyll conductance ( $g_m$ ) are indicated. Revised from Flexas et al. [39].

with the oxygen level determines the relative Rubisco activity for CO<sub>2</sub> fixation,  $g_m$  has to be incorporated into the FvCB model. A number of methods have been developed to estimate  $g_m$  [40–42]. The estimated  $g_m$  appears to respond to temperature [6,43,44] and this response may be described by Eq. (6) [6]. It has been shown recently that  $g_m$  may also vary with CO<sub>2</sub> and irradiance [45,39,20], but this variation is not yet completely certain (see review of [41] and references therein). Thus, we currently assume that  $g_m$  does not vary with CO<sub>2</sub> or with irradiance.

A coupled modelling of leaf photosynthesis and stomatal conductance has been reported frequently in the literature (e.g., [16,46]). Few whole-plant modelling studies have considered  $g_m$  as a necessary term of the photosynthesis models [39]. Here we incorporate  $g_b$  (boundary-layer conductance),  $g_s$  and  $g_m$  into our modelling framework. First, on the basis of several existing models, we propose the following phenomenological sub-model for  $g_s$ :

$$g_s = g_0 + \frac{A + R_d}{C_i - C_{i*}} f_{vpd} \quad (15)$$

where  $g_0$  is the residual stomatal conductance if the irradiance approaches zero,  $C_{i*}$  is the C<sub>i</sub>-based CO<sub>2</sub> compensation point in the absence of  $R_d$  (by definition  $C_{i*} = \Gamma^* - R_d/g_m$ ), and  $f_{vpd}$  is the function for the effect of leaf-to-air vapour pressure difference (VPD), which is not yet understood sufficiently and may be described empirically as:

$$f_{vpd} = \frac{1}{[1/(a_1 - b_1 VPD) - 1]} \quad (15a)$$

where  $a_1$  and  $b_1$  are empirical constants. Eq. (15) is consistent with the finding that stomata may sense  $C_i$  [47]. Furthermore, unlike the model of Leuning et al. [16], Eq. (15) uses  $(A + R_d)$  instead of  $A$  to avoid a possible negative  $g_s$  below the light compensation point. Unlike the model of Dewar [48], Eq. (15) predicts a non-zero  $g_s$  if  $C_i = C_{i*}$ . It also differs from the model of Tuzet et al. [46] in that there is no need to calculate  $\Gamma$ —the CO<sub>2</sub> compensation point in the presence of  $R_d$ .

The following equations can be written, according to Fick's first law of diffusion for CO<sub>2</sub> transfer along the path from  $C_a$  to  $C_c$ :

$$C_i = C_a - A \left( \frac{1}{g_b} + \frac{1}{g_s} \right) \quad (16)$$

$$C_c = C_i - \frac{A}{g_m} \quad (17)$$

The Rubisco-limited ( $A_c$ ) and e<sup>-</sup> transport-limited ( $A_j$ ) parts of the FvCB model can be written in a single equation as:

$$A = \frac{(C_c - \Gamma^*)x_1}{C_c + x_2} - R_d \quad (18)$$

where for the Rubisco-limited part  $x_1 = V_{cmax}$  and  $x_2 = K_{mC}(1 + O/K_{mO})$ , and for the e<sup>-</sup> transport-limited part  $x_1 = J/4$  and  $x_2 = 2\Gamma^*$ .

Solving  $A$  from Eqs. (15)–(18) is commonly done by a numerical iteration approach (e.g., [16]). From applications in a large simulation framework such as for crop growth modelling, a faster approach is preferred. Here, an analytical approach is applied by combining and manipulating Eqs. (15)–(18) into the form of a standard cubic equation for  $A$ :

$$A^3 + pA^2 + qA + r = 0 \quad (19)$$

The analytical solution for a general cubic equation is given in Appendix A. Lumped coefficients  $p$ ,  $q$ , and  $r$  in Eq. (19) are given in Appendix B. The root  $A_1$  in Appendix A was found to be suitable for calculating either  $A_c$  or  $A_j$  under any combinations of  $C_a$ ,  $I_{inc}$ , temperature, and VPD. The minimum of  $A_c$  or  $A_j$  gives  $A$  according to Eq. (1).

## 5. The C<sub>4</sub> model

In C<sub>4</sub> plants, CO<sub>2</sub> is fixed initially in the mesophyll by phosphoenolpyruvate (PEP) carboxylase into C<sub>4</sub> acids that are then decarboxylated to supply CO<sub>2</sub> to Rubisco, which is localized in the bundle-sheath chloroplasts (Fig. 5). The well co-ordinated functioning of mesophyll and bundle-sheath cells, accomplished through specialized leaf anatomy, produces a high CO<sub>2</sub> concentration in the bundle sheath, strongly inhibiting photorespiration. However, the elevated CO<sub>2</sub> in the bundle-sheath cells is sustained at the cost of extra ATP, required for the regeneration of PEP.

The conductance for CO<sub>2</sub> transfer from intercellular air spaces to mesophyll cells may be large enough in C<sub>4</sub> leaves [49]. However, the bundle-sheath conductance ( $g_{bs}$ ) is a major factor that determines the rate of CO<sub>2</sub> leakage from the bundle sheath to the mesophyll ( $L$ ), and  $g_{bs}$  should be small enough for concentrating CO<sub>2</sub> in the bundle sheath. Following the model of von Caemmerer and Furbank [9], which was built upon several earlier models (e.g., [50]), the following two equations specific for C<sub>4</sub> photosynthesis can be written:

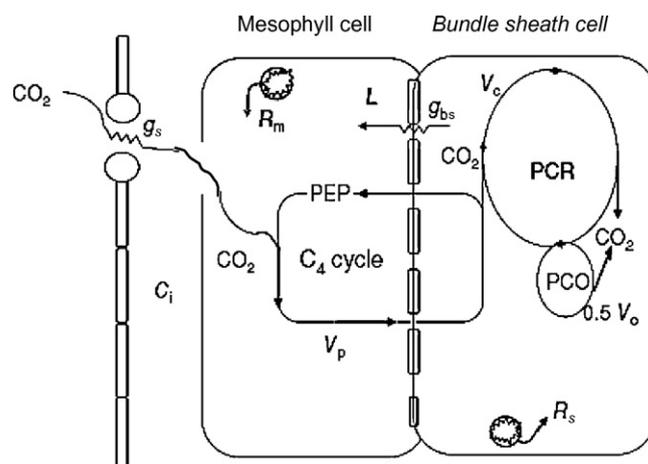
$$L = g_{bs}(C_c - C_i) \quad (20a)$$

$$A = V_p - L - R_m \quad (20b)$$

where  $V_p$  is the rate of PEP carboxylation, and  $R_m$  is the mitochondrial respiration occurring in the mesophyll, which for practical purposes can be set to  $0.5R_d$ .  $V_p$  can be limited either by the activity of PEP carboxylase or by the rate of e<sup>-</sup> transport. For the enzyme-limited case, von Caemmerer and Furbank [9] used a Michaelis–Menten equation to describe  $V_p$ . In order to find an analytical solution when combined with a  $g_s$  model, we use the version of Collatz et al. [51] and He and Edwards [52]:

$$V_p = \min(k_p C_i, V_{pmax}) \quad (20c)$$

where  $k_p$  is the initial carboxylation efficiency of the PEP carboxylase, and  $V_{pmax}$  is the maximum rate of PEP carboxylation at the saturated  $C_i$ , the temperature response of which can be described by Eq. (6) [53]. Use of Eq. (20c) simplifies the model solution, but the



**Fig. 5.** Scheme of the C<sub>4</sub> photosynthesis model based on von Caemmerer and Furbank [9]. After passing the stomatal conductance ( $g_s$ ) and entering the mesophyll cell, CO<sub>2</sub> is fixed by phosphoenolpyruvate (PEP) carboxylase at the rate of  $V_p$ . The formed C<sub>4</sub> acid crosses a bundle-sheath conductance ( $g_{bs}$ ) and is decarboxylated at the same rate  $V_p$ . The released CO<sub>2</sub> either leaks back to the mesophyll cell ( $L$ ) or can be fixed at the bundle-sheath cell by Rubisco at the rate  $V_c$  in the photosynthetic carbon reduction cycle (PCR – the normal C<sub>3</sub> cycle). Part of the CO<sub>2</sub> is again released by the photosynthetic carbon oxidation (PCO) cycle at half the rate of Rubisco oxygenation ( $V_o$ ). CO<sub>2</sub> can also be released in the mesophyll and bundle sheath from mitochondrial respiration ( $R_m$  and  $R_s$ ), which together make the total day respiration rate  $R_d$ .

modelled CO<sub>2</sub> response curve may reach saturation a little sooner than that given by using a Michaelis–Menten equation.

For the e<sup>-</sup> transport-limited V<sub>p</sub>, we use a generalized form (unpublished):

$$V_p(J_2) = \frac{xJ_2(2 + f_Q - f_{cyc})}{2h(1 - f_{cyc})} = \frac{xJ_2z}{2} \quad (20d)$$

where  $z = (2 + f_Q - f_{cyc})/[h(1 - f_{cyc})]$ ;  $x$  represents the fraction of the total PSII e<sup>-</sup> transport rate ( $J_2$ ) partitioned to the mesophyll reactions, so  $(1 - x)$  is the fraction of  $J_2$  partitioned to the bundle-sheath reactions; the definition of the other variables in Eq. (20d) is the same as used for the C<sub>3</sub> generalized model, Eqs. (10a)–(10d).

Eqs. (20a) and (20b) can be combined to result in:

$$C_c = C_i + \frac{V_p - A - R_m}{g_{bs}} \quad (21)$$

Incorporating algorithms from Eqs. (20c) and (20d), Eq. (21) can have different forms, depending on how V<sub>p</sub> is calculated:

$$C_c = aC_i + \frac{b - A - R_m}{g_{bs}} \quad (22)$$

$$\text{where } a = \begin{cases} 1 + k_p/g_{bs} & \text{if } V_p = k_p C_i \\ 1 & \text{if } V_p = V_{p\max} \text{ or } V_p(J_2) \end{cases}$$

$$\text{and } b = \begin{cases} 0 & \text{if } V_p = k_p C_i \\ V_{p\max} & \text{if } V_p = V_{p\max} \\ V_p(J_2) & \text{if } V_p = V_p(J_2) \end{cases}$$

As in C<sub>3</sub> photosynthesis, the rate of CO<sub>2</sub> fixation by Rubisco in C<sub>4</sub> photosynthesis can be limited either by the Rubisco activity (see Eq. (2)) or by the e<sup>-</sup> transport. In combination, the net rate of CO<sub>2</sub> assimilation can be expressed as:

$$A = \frac{(C_c - \gamma^* O_{bs})x_1}{C_c + x_2 O_{bs} + x_3} - R_d \quad (23)$$

where  $O_{bs}$  is the O<sub>2</sub> concentration in the bundle-sheath cell;  $\gamma^* = 0.5/S_{c/o}$ ;  $x_1 = V_{c\max}$ ,  $x_2 = K_{mC}/K_{mO}$ ,  $x_3 = K_{mC}$  for the enzyme (Rubisco)-limited rate, and  $x_1 = (1 - x)J_2z/3$ ,  $x_2 = 7\gamma^*/3$  and  $x_3 = 0$  for the e<sup>-</sup> transport-limited rate (unpublished). This form of the e<sup>-</sup> transport-limited rate implicitly assumes that it is the ATP supply rather than the NADPH supply that causes the e<sup>-</sup> transport limitation in C<sub>4</sub> photosynthesis as a whole. This assumption holds because there is no net NADPH requirement by the C<sub>4</sub> cycle itself although in NADP-ME C<sub>4</sub> subtype, NADPH consumed in the production of malate is released in the bundle sheath during decarboxylation. However, there is the additional cost of 2 mol of ATP for the regeneration of 1 mol of PEP from pyruvate in the mesophyll [9].

In addition, PSII activity and O<sub>2</sub> evolution in the bundle sheath vary widely amongst the C<sub>4</sub> species. This has implications for the steady-state O<sub>2</sub> partial pressure of the bundle sheath. Following von Caemmerer and Furbank [9], a relation between intercellular air-space O<sub>2</sub> partial pressure ( $O_i$ ) and the bundle-sheath O<sub>2</sub> partial pressure ( $O_{bs}$ ) is described as:

$$O_{bs} = \frac{\alpha A}{0.047g_{bs}} + O_i \quad (24)$$

where  $\alpha$  is the fraction of O<sub>2</sub> evolution occurring in the bundle sheath, and 0.047 accounts for the diffusivities for O<sub>2</sub> and CO<sub>2</sub> in water and their respective Henry constants [9]. For maize and sorghum,  $\alpha$  will be zero whereas it will approach or even exceed 0.5 in other cases [54].

CO<sub>2</sub> transfer along the path from C<sub>a</sub> to C<sub>i</sub> is the same as in Eq. (16) for C<sub>3</sub> photosynthesis. However, to permit finding an analyti-

cal solution for A in C<sub>4</sub> photosynthesis, we use a slightly different phenomenological equation as a g<sub>s</sub> model:

$$g_s = g_0 + \frac{A + R_d}{C_s - C_{s^*}} f_{vpd} \quad (25)$$

where C<sub>s</sub> is the CO<sub>2</sub> level at leaf surface, and C<sub>s\*</sub> is the C<sub>s</sub>-based CO<sub>2</sub> compensation point in the absence of R<sub>d</sub>. By definition, C<sub>s\*</sub> and C<sub>i\*</sub> differ by  $C_{s^*} = C_{i^*} - R_d/g_{s,C_{i^*}}$ , where  $g_{s,C_{i^*}}$  is the stomatal conductance at C<sub>i\*</sub>. The value of C<sub>i\*</sub> can be solved from Eqs. (21) and (24) with  $C_c = \gamma^* O_{bs}$ ,  $V_p = k_p C_{i^*}$  and  $A = -R_d$ :

$$C_{i^*} = \frac{g_{bs}\gamma^* O_i - (1 + \gamma^*\alpha/0.047)R_d + R_m}{g_{bs} + k_p} \quad (26)$$

Combining Eqs. (16) and (22)–(25) and manipulating them can yield a form of cubic equation, Eq. (19), in which its coefficients are given in Appendix C. The root A<sub>3</sub> in Appendix A was found to be suitable for calculating A under any combination of C<sub>a</sub>, J<sub>inc</sub>, temperature, and VPD. As either the enzyme activity or the e<sup>-</sup> transport can limit both Rubisco and PEP-carboxylase reactions, in theory four types of combinations of rate limitations are possible. The minimum of the four rates gives the prediction of A.

There is not published a simple calibration procedure as yet to account for alternative e<sup>-</sup> transports and other uncertain factors as done for C<sub>3</sub> photosynthesis, see Eqs. (11)–(13). For C<sub>4</sub> photosynthesis, however, it is certain that the ATP supply causes the e<sup>-</sup> transport limitation. Although non-chloroplastic processes (e.g., oxidative phosphorylation in mitochondria) could provide additional ATP in the PCK C<sub>4</sub> subtype, it is most likely that the high ATP requirement is largely met within the chloroplasts in the main C<sub>4</sub> crop species (maize and sorghum), since the respiration cost in C<sub>4</sub> leaves is generally not higher than in C<sub>3</sub> leaves [55]. An equation equivalent to Eq. (10d) for C<sub>3</sub> photosynthesis, can be derived for C<sub>4</sub> photosynthesis (unpublished):

$$1 - f_{cyc} - f_{pseudo} = \frac{(4C_c + 8\gamma^* O_{bs})(2 + f_Q - f_{cyc})(1 - x)}{h(3C_c + 7\gamma^* O_{bs})} \quad (27)$$

Comparing Eq. (27) with Eq. (10d) shows that if f<sub>Q</sub> and h were the same for C<sub>3</sub> and C<sub>4</sub> photosynthesis, the fraction for LET, i.e.,  $(1 - f_{cyc} - f_{pseudo})$  (see Fig. 3), would need to be decreased in C<sub>4</sub> photosynthesis by a factor  $(1 - x)$ . To meet the high demand for ATP in C<sub>4</sub> photosynthesis, however, the Q-cycle, an efficient proton-pumping routine along the LET chain (Fig. 3), is obligatory [56]. So f<sub>Q</sub> is set to 1. As there are minor physiological processes simultaneously occurring with CO<sub>2</sub> fixation that consume e<sup>-</sup> or reductants from the chloroplasts [57,58], the value of f<sub>pseudo</sub> can be practically assigned, e.g., ca. 0.1 for both C<sub>3</sub> [35] and C<sub>4</sub> plants [59], but with caution as the exact value of f<sub>pseudo</sub> is unknown. These minor physiological processes are beyond the scope of modelling photosynthesis in this paper and are not described in detail. Assuming photorespiration is negligible in C<sub>4</sub> leaves, the required f<sub>cyc</sub> can be solved from Eq. (27) as:

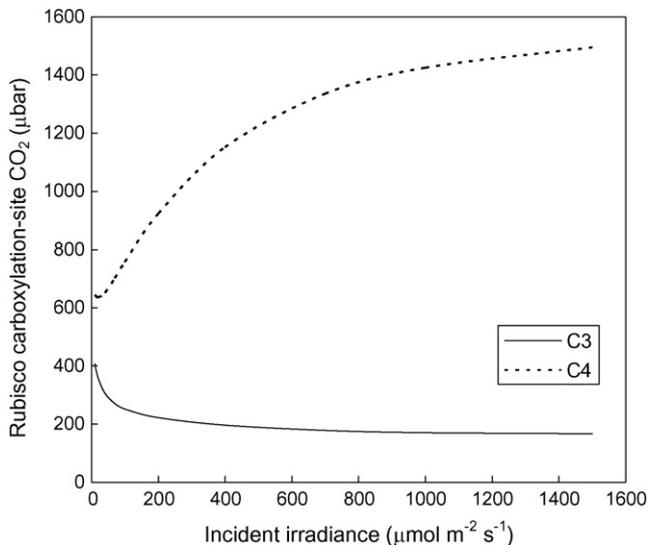
$$f_{cyc} = 1 - \frac{4(1 - x)(1 + f_Q) + 3hf_{pseudo}}{3h - 4(1 - x)} \quad (28)$$

where x may be set at 0.4, based on an optimization analysis for a wide range of conditions ([9]; and references therein). There is uncertainty about the value of parameter h (3, 4 or 14/3), which has been debated for C<sub>3</sub> photosynthesis [35,54]. Eq. (28) predicts that f<sub>cyc</sub> is 0.136, 0.375 and 0.466 if h is 3, 4 and 14/3, respectively. When f<sub>cyc</sub> is known, the required J<sub>2</sub>, see Eqs. (20d) and (23), can be calculated using Eqs. (10b) and (10c). Although simulation shows that C<sub>c</sub> can be low enough (Fig. 6) for small photorespiration to occur at limiting light levels, simulated values for  $\Phi_{CO_2(LL)}$  using the above assumptions agree with the range of measured  $\Phi_{CO_2(LL)}$  for C<sub>4</sub> species [37,60,61], with h = 3 giving  $\Phi_{CO_2(LL)}$  at the upper side and h = 14/3 giving a value at the lower side of the range. However,

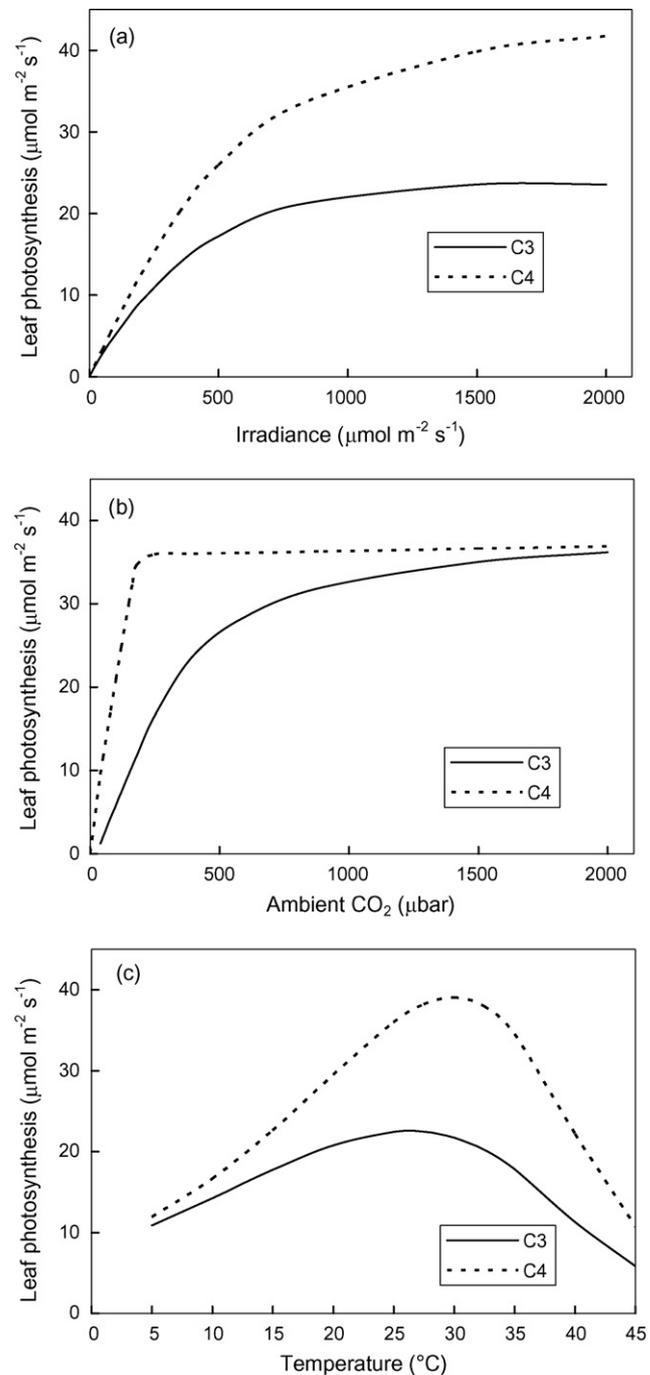
the reported interspecific difference in  $\Phi_{\text{CO}_2(\text{LL})}$  could be due to the variation in other parameters (e.g.,  $g_{\text{bs}}$ ). Furthermore, it is unclear whether the chloroplasts are flexible enough (e.g., via state transition) to support CET with a high  $f_{\text{cyc}}$  close to or even above 0.4. Thus, for the following section we use the conservative value for  $h$  (i.e., 3), which was previously assumed to assess quantum requirement of  $\text{C}_4$  photosynthesis (e.g., [50,56]), but with the caution that  $h$  is likely to be 4.

## 6. Simulation highlights and some remarks from the perspective of crop modelling

Some constants characterizing temperature response of enzyme kinetics are not yet well quantified for  $\text{C}_4$  photosynthesis (Table 2). However, in addition to those for  $\text{C}_3$  photosynthesis, the most important algorithms for  $\text{C}_4$  photosynthesis are the ones that capture the  $\text{CO}_2$  concentrating mechanism and the extra ATP requirement by the PEP regeneration. Fig. 6 compares the level of the simulated  $C_c$  in  $\text{C}_3$  and  $\text{C}_4$  leaves as a function of irradiance. At high light levels,  $C_c$  of  $\text{C}_4$  leaves is 8–10 times that of  $\text{C}_3$  leaves, which largely explains why there is virtually no photorespiration in  $\text{C}_4$  leaves. At low light levels, the difference is smaller, due to a higher fraction of leakage of  $\text{CO}_2$  back to the mesophyll cell. As a result, the predicted light response curve saturates at a much higher light level, the predicted  $\text{CO}_2$  response curve saturates at a much lower  $\text{CO}_2$  level, and the predicted temperature response curve has a higher temperature optimum in  $\text{C}_4$  than  $\text{C}_3$  photosynthesis (Fig. 7), whereas the predicted difference in the photosynthetic quantum efficiency  $\Phi_{\text{CO}_2(\text{LL})}$  at a reference  $25^\circ\text{C}$  is relatively small (Fig. 8). We emphasize that the modelled differences in response curves and values of  $\Phi_{\text{CO}_2(\text{LL})}$  between  $\text{C}_3$  and  $\text{C}_4$  cases should not be considered as absolute as we only used default parameter values to illustrate general trends. Overall,  $\text{C}_4$  photosynthesis response curves are very similar to those for  $\text{C}_3$  photosynthesis measured under low  $\text{O}_2$  or/and high  $\text{CO}_2$  conditions (which can suppress



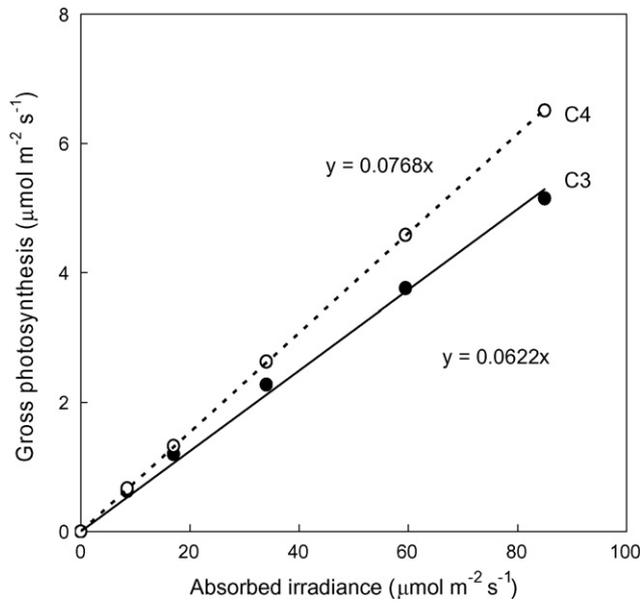
**Fig. 6.** Comparison of simulated  $\text{CO}_2$  partial pressure at the Rubisco carboxylation site between  $\text{C}_3$  and  $\text{C}_4$  leaves. For this simulation,  $C_a = 360 \mu\text{bar}$ ,  $T = 25^\circ\text{C}$ ,  $\text{VPD} = 2.1 \text{ kPa}$ ,  $O_i = 210,000 \mu\text{bar}$ ,  $V_{\text{cmax}25} = 120 \mu\text{mol m}^{-2} \text{ s}^{-1}$ ,  $J_{\text{max}25} = 230 \mu\text{mol m}^{-2} \text{ s}^{-1}$ ,  $R_{\text{d}25} = 0.01 V_{\text{cmax}25}$ ,  $g_b = 1.5 \text{ mol m}^{-2} \text{ s}^{-1} \text{ bar}^{-1}$ ,  $g_o = 0.01 \text{ mol m}^{-2} \text{ s}^{-1} \text{ bar}^{-1}$ ,  $a_1 = 0.9$ ,  $b_1 = 0.15 \text{ kPa}^{-1}$ . For  $\text{C}_3$ :  $g_m = 0.4 \text{ mol m}^{-2} \text{ s}^{-1} \text{ bar}^{-1}$ ; for  $\text{C}_4$ :  $\alpha = 0.1$ ;  $g_{\text{bs}} = 0.003 \text{ mol m}^{-2} \text{ s}^{-1} \text{ bar}^{-1}$ . For  $\text{C}_4$  simulation, the small difference between  $C_s^-$  and  $C_i^-$  was assumed negligible; furthermore,  $V_{\text{pmax}}$  was tentatively assumed not limiting on  $V_p$  because how it scales quantitatively with  $V_{\text{cmax}}$  or with  $J_{\text{max}}$  is not yet very clear; but this assumption has very little impact on the general conclusions of this and subsequent simulations because with increasing  $C_i$ ,  $V_p(J_2)$  will soon replace  $k_p C_i$  to become limiting on  $V_p$ .



**Fig. 7.** Simulated curves of  $\text{C}_3$  and  $\text{C}_4$  gross photosynthesis in response to irradiance where  $C_a = 360 \mu\text{bar}$  and  $T = 25^\circ\text{C}$  (a), to ambient  $\text{CO}_2$  where  $I_{\text{inc}} = 1000 \mu\text{mol m}^{-2} \text{ s}^{-1}$  and  $T = 25^\circ\text{C}$  (b), and to temperatures where  $I_{\text{inc}} = 1000 \mu\text{mol m}^{-2} \text{ s}^{-1}$  and  $C_a = 360 \mu\text{bar}$  (c). Model constants used for this simulation are shown in Table 2; for other parameters see Fig. 6.

photorespiration), except that  $\Phi_{\text{CO}_2(\text{LL})}$  in  $\text{C}_4$  leaves is significantly lower than  $\Phi_{\text{CO}_2(\text{LL})}$  in  $\text{C}_3$  leaves under the non-photorespiratory condition, the latter having a value of 0.09–0.11 [37,62].

As stated earlier,  $\Phi_{\text{CO}_2(\text{LL})}$  is an important parameter for the photosynthetic light response curve in many crop growth models. Our model algorithms predict that at a  $C_a$  of  $360 \mu\text{bar}$ ,  $\Phi_{\text{CO}_2(\text{LL})}$  decreases sharply with increasing temperature in  $\text{C}_3$  plants but remains virtually insensitive to temperature in  $\text{C}_4$  plants (Fig. 9a), consistent with experimental evidence [60,61]. The predicted sensitivity of  $\Phi_{\text{CO}_2(\text{LL})}$  to temperature for  $\text{C}_3$  plants at a doubled  $\text{CO}_2$



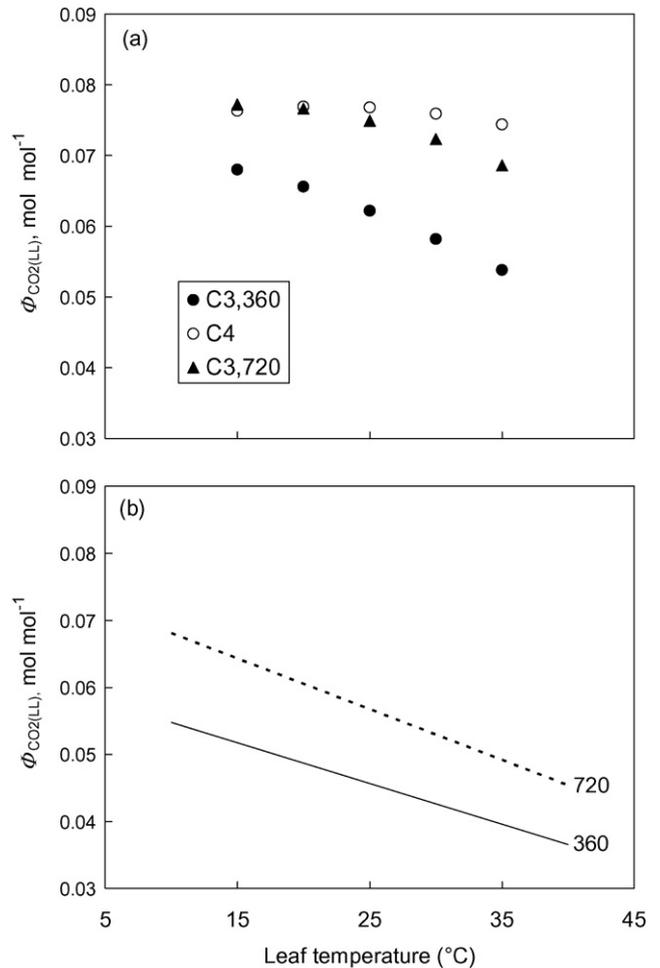
**Fig. 8.** Illustration of the calculation of photosynthetic quantum efficiency at limiting light levels ( $\Phi_{\text{CO}_2(\text{LL})}$ ) using simulated gross  $\text{CO}_2$  assimilation rates  $A + R_d$  (points).  $C_a = 360 \mu\text{bar}$  and  $T = 25^\circ\text{C}$ . The slope of the linear regression through the origin is  $\Phi_{\text{CO}_2(\text{LL})}$ . Model constants used for this simulation are shown in Table 2; for other parameters see Fig. 6.

level is reduced, compared with the response under the ambient  $\text{CO}_2$  condition (Fig. 9a). This is supported by the experimental measurement of Ku and Edwards [63] that the sensitivity of  $\Phi_{\text{CO}_2(\text{LL})}$  to temperature in wheat is reduced by a reduced  $\text{O}_2$  level. To quantify the interaction between the effect of temperature,  $f(T)$ , and the effect of  $C_a$ ,  $g(C_a)$ , on  $\Phi_{\text{CO}_2(\text{LL})}$ , the rice growth model ORYZA [31] uses the following empirical, multiplicative expression:

$$\Phi_{\text{CO}_2(\text{LL})} = \Phi_{\text{CO}_2(\text{LL})\text{ref}}(T)g(C_a) \quad (29)$$

where  $\Phi_{\text{CO}_2(\text{LL})\text{ref}}$  is  $\Phi_{\text{CO}_2(\text{LL})}$  at a reference temperature and  $C_a$ ; and  $f(T)$  and  $g(C_a)$  were established from separate experimental data. This model predicts an increased temperature sensitivity of  $\Phi_{\text{CO}_2(\text{LL})}$  under the elevated  $\text{CO}_2$  condition (Fig. 9b). The earlier version of the ORYZA model was used to assess the impact of global warming on the Asian rice production [64]. It is clear that its simulation results should receive a critical reservation because the underlying sub-model for photosynthesis does not predict correctly (1) the temperature optimum shift and the temperature range (within which  $A$  is positive) at elevated  $\text{CO}_2$  (Fig. 2) and (2) the direction of interaction between temperature and elevated  $\text{CO}_2$  on  $\Phi_{\text{CO}_2(\text{LL})}$  (Fig. 9).

Multiplicative models like Eq. (29) have often been used by crop modellers as a standard method to describe how physiological processes respond to two or more interacting variables. The example shown in Fig. 9 means that this form can be incorrect. Algorithms in major crop growth models have been updated little since the late 1980s, indicating “a sense of arrogance and complacency” [65]. To face new challenges in crop science, crop modellers should be willing to utilize the rich knowledge at a lower scale, such as elegant yet simple FvCB-type photosynthesis models already available for years. A common view of crop modellers is that parameterization of the FvCB model for different crops is difficult and time-consuming [66]. Given the increasing availability of a wealth of information for the key enzyme constants (Table 2), which are believed to be conservative among  $C_3$  or  $C_4$  species, the task of parameterization can focus on a few key parameters, estimated by curve fitting to readily available gas exchange measurements (e.g., [17,19]). The version of  $C_3$  and  $C_4$  photosynthesis models presented here has just



**Fig. 9.** Temperature response of photosynthetic quantum efficiency under limiting lights  $\Phi_{\text{CO}_2(\text{LL})}$ , simulated by the  $C_3$  and  $C_4$  models described here (a), or by the empirical multiplicative Eq. (29) as used in the ORYZA model [31] (b). For the  $C_3$  case in both panels, two levels of  $C_a$  (360 and 720  $\mu\text{bar}$ ) were considered, whereas for the  $C_4$  case shown in panel (a),  $C_a = 360 \mu\text{bar}$ . Model constants used for simulation shown in panel (a) are given in Table 2; for other parameters see Fig. 6.

been incorporated into a relatively new crop growth model GECROS [24]. This model allows one to objectively examine a number of important research questions, such as exploring options for bio-energy production and assessing impacts of global warming and transformation of  $C_4$  routine into  $C_3$  crops on season-long canopy photosynthesis and crop grain yields. Crop modelling has been by and large reliant on simple approaches; however, in the words attributed to Albert Einstein, “Everything should be made as simple as possible, but not simpler”.

#### Appendix A. Analytical solution of a cubic equation—Eq. (19)

The solution of the standard form of a cubic equation like Eq. (19) is taken from Press et al. [67]; see also Collatz et al. [51] and Baldocchi [68]. Three roots for the equation are:

$$A_1 = -2\sqrt{Q} \cos\left(\frac{\psi}{3}\right) - \frac{p}{3}$$

$$A_2 = -2\sqrt{Q} \cos\left(\frac{\psi + 2\pi}{3}\right) - \frac{p}{3}$$

$$A_3 = -2\sqrt{Q} \cos\left(\frac{\psi + 4\pi}{3}\right) - \frac{p}{3}$$

$$\text{where } Q = (p^2 - 3q)/9; \quad \psi = \arccos(U/\sqrt{Q^3}); \\ U = (2p^3 - 9pq + 27r)/54.$$

### Appendix B. Lumped coefficients in Eq. (19) for the coupled C<sub>3</sub> photosynthesis and diffusional conductance model

The coefficients  $p$ ,  $q$  and  $r$  of Eq. (19) for C<sub>3</sub> photosynthesis are:

$$p = -\frac{[d + (x_1 - R_d)/g_m + a(1/g_m + 1/g_b) + (g_0/g_m + f_{vpd})c]}{m}$$

$$q = \frac{[d(x_1 - R_d) + ac + (g_0/g_m + f_{vpd})b]}{m}$$

$$r = -\frac{ab}{m}$$

where the coefficients  $a$ ,  $b$ ,  $c$ ,  $d$  and  $m$  are expressed as:

$$a = g_0(x_2 + \Gamma_*) + \left(\frac{g_0}{g_m} + f_{vpd}\right)(x_1 - R_d)$$

$$b = C_a(x_1 - R_d) - \Gamma_*x_1 - R_dx_2$$

$$c = C_a + x_2 + \left(\frac{1}{g_m} + \frac{1}{g_b}\right)(x_1 - R_d)$$

$$d = x_2 + \Gamma_* + \frac{(x_1 - R_d)}{g_m}$$

$$m = \frac{1}{g_m} + \left(\frac{g_0}{g_m} + f_{vpd}\right)\left(\frac{1}{g_m} + \frac{1}{g_b}\right)$$

where  $x_1$  and  $x_2$  are defined in the texts following Eq. (18).

### Appendix C. Lumped coefficients in Eq. (19) for the coupled C<sub>4</sub> photosynthesis and diffusional conductance model

The coefficients  $p$ ,  $q$  and  $r$  of Eq. (19) for C<sub>4</sub> photosynthesis are:

$$p = \frac{[j - (h - lR_d)]}{l}$$

$$q = \frac{(i + jR_d - g)}{l}$$

$$r = -\frac{(f - iR_d)}{l}$$

where the coefficients  $f$ ,  $g$ ,  $h$ ,  $i$ ,  $j$  and  $l$  are expressed as:

$$f = (b - R_m - \gamma_*O_i g_{bs})x_1 d + ag_{bs}x_1 C_a d$$

$$g = (b - R_m - \gamma_*O_i g_{bs})x_1 m - \left(\frac{\alpha\gamma_*}{0.047} + 1\right)x_1 d \\ + ag_{bs}x_1 \left[C_a m - \frac{d}{g_b} - (C_a - C_{s*})\right]$$

$$h = -\left[\left(\frac{\alpha\gamma_*}{0.047} + 1\right)x_1 m + \frac{ag_{bs}x_1(m-1)}{g_b}\right]$$

$$i = (b - R_m + g_{bs}x_3 + x_2 g_{bs}O_i)d + ag_{bs}C_a d$$

$$j = (b - R_m + g_{bs}x_3 + x_2 g_{bs}O_i)m + \left(\frac{\alpha x_2}{0.047} - 1\right)d \\ + ag_{bs} \left[C_a m - \frac{d}{g_b} - (C_a - C_{s*})\right]$$

$$l = \left(\frac{\alpha x_2}{0.047} - 1\right)m - \frac{ag_{bs}(m-1)}{g_b}$$

where  $x_1$ ,  $x_2$  and  $x_3$  are defined in the texts following Eq. (23),  $a$  and  $b$  are defined in the equations below Eq. (22), and  $d$  and  $m$  are defined as:

$$d = g_0 C_a - g_0 C_{s*} + f_{vpd} R_d$$

$$m = f_{vpd} - \frac{g_0}{g_b}$$

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